**Serial Dilution Experiment Sheet**

Vinegar Serial Dilution

Materials:

5 clear glass containers

graduated cylinder

water

food coloring

Alka-Seltzer tablets

stopwatch

distilled white vinegar

Procedure:

1. Measure 100 mL of water and pour into one of the clear containers.
2. Add one drop of food coloring. Stir. (Blue makes the best contrast for visual acuity).
3. Add one Alka-Seltzer tablet, and use a stopwatch to measure how long it takes for the tablet to dissolve.
4. Record the time.
5. Create a data table with columns for concentration and time.
6. Repeat the above procedure using 100 mL of distilled white vinegar.
7. Repeat the procedures using the following solutions (1 solution per container): The 25% solution consists of 25 mL of distilled white vinegar and 75 mL of water, the 50% solution consists of 50 mL of distilled white vinegar and 50 mL of water, and the 75% solution consists of 75mL of distilled white vinegar and 25 mL of water.

**Data Table**

|  |  |  |  |  |
| --- | --- | --- | --- | --- |
| **Water** | **Vinegar** | **Solution Volume** | **Percent Solution** | **Time to dissolve**  |
| 100mL |  | 100mL |  |  |
|  | 25mL |  |  |  |
| 50mL |  |  |  |  |
|  | 75mL |  |  |  |
|  |  |  |  |  |

Conclusion:

1. Explain how the changes in concentration changed the amount of time it took the Alka-Seltzer to dissolve.
2. How can you use this information when engineering your dye concentrations?