

Viscosity Activity Worksheet

1. Describe the fluid you are working with using every day descriptive vocabulary. (For example: “I am looking at honey. It is yellow(ish) and clear(ish). It is pretty thick and moves slowly. It feels sticky.”)

2. Calculate the density of the fluid using these steps:

- Weigh the empty graduated cylinder. Record its mass in grams.

$$M_{\text{cylinder}} = \text{_____} \text{ [g]}$$

- Fill the cylinder with fluid, and record the volume in cm^3 . Note: $1 \text{ cm}^3 = 1 \text{ ml}$.

$$\text{Vol}_{\text{fluid}} = \text{_____} \text{ [cm}^3\text{]}$$

- Weigh the full graduated cylinder. Subtract the mass of the empty graduated cylinder and record the mass of the fluid.

$$M_{\text{fluid}} = \text{_____} \text{ [g]}$$

- The density of the fluid is the mass over the volume. Calculate the density of the fluid.

$$\rho_f = \frac{\text{Mass of Fluid [g]}}{\text{Volume of Fluid [cm}^3\text{]}}$$

$$\rho_f = \text{_____} \text{ [g/cm}^3\text{]}$$

Name: _____ Date: _____

3. Measure the density of the sphere using these steps:

- Measure the radius of the sphere. Record as r [cm].

$$r_s = \text{_____ [cm]}$$

- Calculate the volume of the sphere. Either use the equation:
or place the sphere in a graduated cylinder filled with water and record its displacement.

$$Vol_s = \text{_____ [cm}^3\text{]}$$

- Weigh the sphere. Record its mass.

$$M_s = \text{_____ [g]}$$

- Calculate the density of the sphere by dividing its mass by its volume.

$$\rho_s = \text{_____ [g/cm}^3\text{]}$$

4. Measure the terminal velocity of the sphere falling through the fluid using these steps:

- With your stopwatch ready, drop the ball into the fluid.

If the fluid is not very viscous, the ball will fall through it very fast, *so be ready!*

If the fluid is thick enough, then the ball will reach a constant speed.

This is the *terminal velocity*, the point at which the drag on the sphere by the fluid is equal to the force of gravity.

- Measure how fast the ball falls a distance. Record the distance, and the time.

$$\text{distance} = \text{_____ [cm]}$$

$$\text{time} = \text{_____ [s]}$$

- Calculate the velocity, which is the distance divided by the time.

$$V_s = \text{_____ [cm/s]}$$

Name: _____ Date: _____

5.

6. **Using this equation, derived from Stokes' law, calculate the viscosity of your fluid.** Gravity is 981 cm/s^2 . *Be very careful* to show your units and how they cancel out. Your final answer should be in units of $[\text{g}/(\text{cm s})]$.

$$\mu = \left(\frac{2}{9}\right) * \frac{r^2 * g(\rho_s - \rho_f)}{V_s}$$

$\mu =$ _____ $[\text{g}/(\text{cm s})]$

7. **Viscosities are usually recorded in $[\text{Pa s}]$. To convert from $[\text{g}/(\text{cm s})]$ to $[\text{Pa s}]$, simply divide by 10:**



$\mu =$ _____ $[\text{Pa s}]$

8. **Using the internet, look up the viscosities of some common household fluids. Be sure to include units. Do any of the answers surprise you?**

Fluid	Viscosity	Fluid	Viscosity
<i>Example: blood</i>	3×10^{-3} to $4 \times 10^{-3} [\text{Pa s}]$		

Note: In searching the internet, you may find viscosities in a variety of units. Some may be in Poise [P] or Centipoise [cP]. $1 [\text{cP}] = .001 [\text{Pa s}]$. The viscosity of water is $1 [\text{cP}]$. Other fluids may have viscosity in Stokes [St], which is the ratio of the viscosity to the density of the fluid. To convert from Stokes, multiply it by the fluid's density, or find another source! *Hint:* Search for "dynamic viscosity."