**Alloy Advantage Summary Assessment**

**Part A:** On the lines below, define *alloy* in your own words.

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**Part B:** Review the table below. Answer the questions that follow.

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|  | **Description** | **Properties** | **Uses** |
| **Iron (Fe)** | * Element (pure substance)
* Atomic number 26
* Fourth most abundant element of Earth’s crust
 | * Density: 7.8 g/cm3
* Melting point: 1538 oC
* Very reactive
* Rapidly corrodes
* Hard
* Brittle
 | * Make alloys
* Vital to plant and animal life; carries oxygen
 |
| **A709 Steel** | * Alloy (mixture)
* Composed of mostly iron, magnesium, silicon and carbon
 | * Density: 7.9 g/cm3
* Melting point: 1510 oC
* Non-corrosive in most environments
* High strength
* Non-brittle
 | * Bridges
* Buildings (skyscrapers)
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1. How do the properties of iron change once an alloy is created?
2. Why might a materials engineer recommend the use of steel over pure iron in the design of bridges?