

# Thermodynamics worksheet

Name: \_\_\_\_\_ Period: \_\_\_\_\_

Date: \_\_\_\_\_

**Design Need:** The customer needs a thermos designed that is capable of the following:

- holding 200mL of water
- spending less than \$3 for all materials
- least amount of heat loss (smallest decrease in temperature of boiling water after 10 minutes)
- lowest total design cost
- using only the materials listed below
- **Can be built & tested by end of class tomorrow (2 days total)**

**Material Specifications:**

Material	Thermal Conductivity (W/m.K)	Cost (\$)
Water (insulation)	0.60	0.25/ mL
Aluminum Foil	250	0.05/ inch
Cotton Balls	0.03	0.15/ each
Paper Cup	0.05	0.55/ each
Sand	0.25	0.01/ gram
Plastic Cup	0.23	0.15/ each
Styrofoam Cup	0.03	0.85/ each
Paint	<i>radiation heat</i>	0.25/ layer
Foam Insulation	0.03	0.05/ inch
Masking Tape	0.08	0.05/ inch

**Heat Loss Equation:**

$$\text{Fourier's Law: } q = \frac{k \cdot A \cdot (T_H - T_C)}{L}$$

where

**q**  
= heat transferred per unit time (W)

**A** = heat transfer area (m<sup>2</sup>)

**k** = thermal conductivity of the material (W/m.K)

**T<sub>H</sub>** = hot temperature (K)

**T<sub>C</sub>** = cold temperature (K)

**L** = material thickness (m)

**1. Complete the following statements based on the introduction from your teacher**

- According to the Heat Loss Equation the value of **q** should be \_\_\_\_\_ (big or small) if we want less heat loss.
- According to the Heat Loss Equation the value of **q** should be \_\_\_\_\_ (big or small) if we want more heat loss.
- According to the Heat Loss Equation the value of **k** should be \_\_\_\_\_ (big or small) if we want less heat loss.

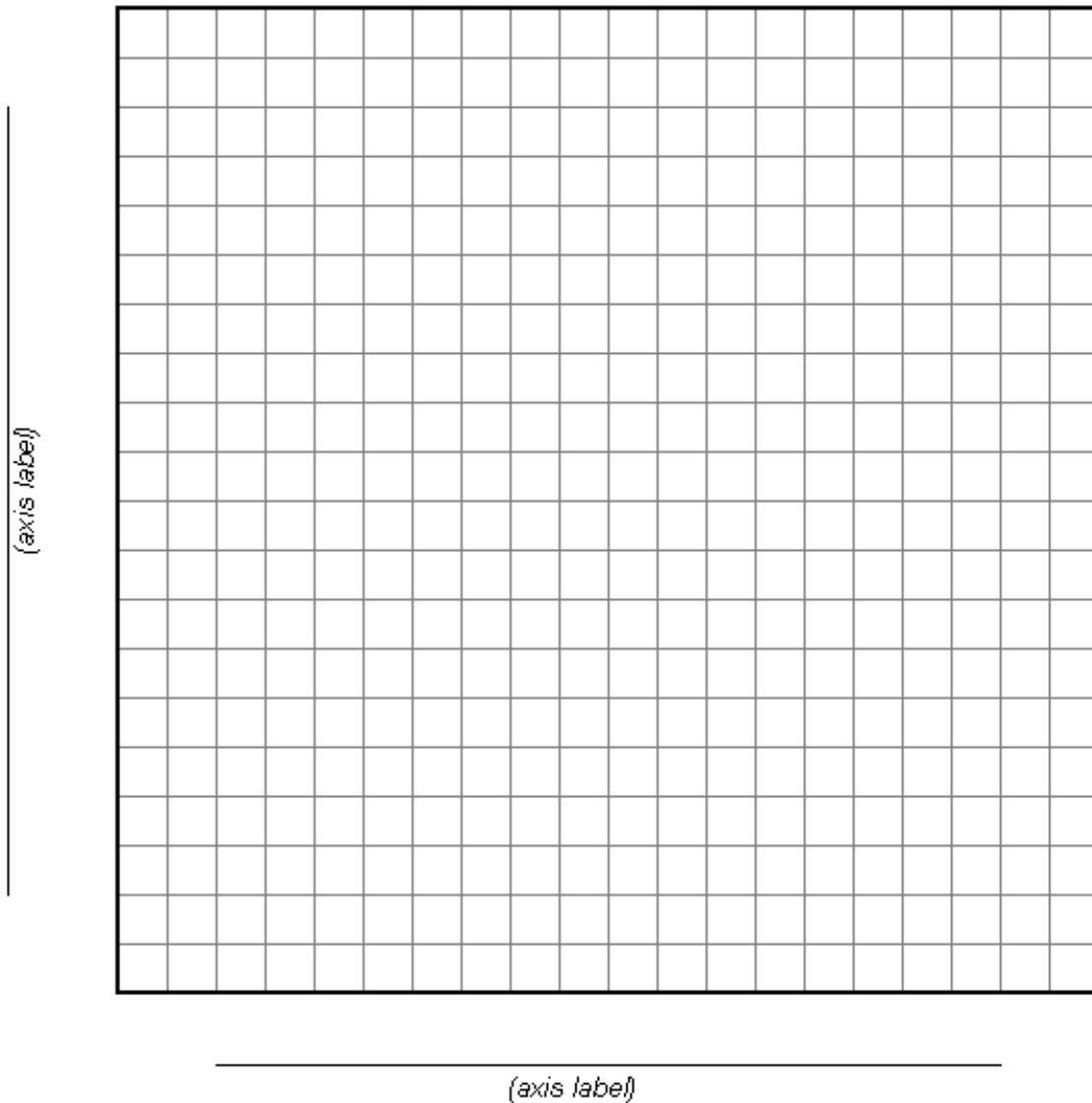
d. According to the Heat Loss Equation the value of **k** should be \_\_\_\_\_ (big or small) if we want more heat loss.





**4. Plot the heat loss of each thermos design**

- y-axis = temperature
- x-axis = time in minutes



**5. What was your best design (#1 or #2)?**

DESIGN COST = \$ \_\_\_\_\_

$\Delta T$  (TEMPERATURE LOSS) = \_\_\_\_\_ °F after 10 minutes

$\$/\Delta T$  = \_\_\_\_\_

**Be ready to present your design to the class.**

